

MODEL STUDIES TOWARDS THE SYNTHESIS OF 2- β -HYDROXYMETHYL PENICILLIN N

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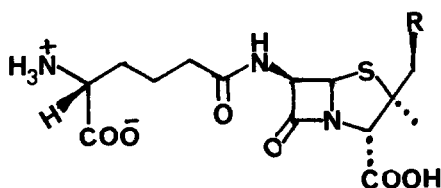
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Abstract: The synthesis of 2- β -hydroxymethyl penicillin V sulfoxide is described as a model study towards the synthesis of 2- β -hydroxymethyl penicillin N.

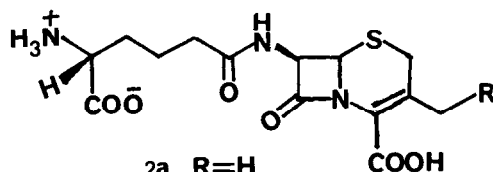
As part of our interest in 2- β -hydroxymethyl penicillin N 1b¹ as a possible intermediate in the cell-free ring expansion of penicillin N 1a into desacetoxycephalosporin C 2a^{2,3,4} we report here the synthetic attempts towards the title compound.

The only report of 2- β -hydroxymethyl penicillin is by Cooper⁵ when he isolated 2- β -hydroxymethyl penicillin V sulfoxide 4 as one of a "plethora of products" by peracidic oxidation of a thiazolidine derivative 3. The approach followed in our studies stems from the observation made by Kamiya⁶ of isolation of penicillin and cephalosporin derivatives when the disulfide 5 was treated with Ag⁺ in the presence of a nucleophile. Thus the disulfide 5, prepared by Kamiya's⁷ procedure, when treated with the silver salt of a carboxylic acid⁸ in the presence of a large excess of the corresponding carboxylic acid gave penicillin derivatives (6a-c) and cephalosporin derivatives (7a-c). The results are summarised in Table 1 and the chemical shifts of various characteristic protons are given in Table 2.

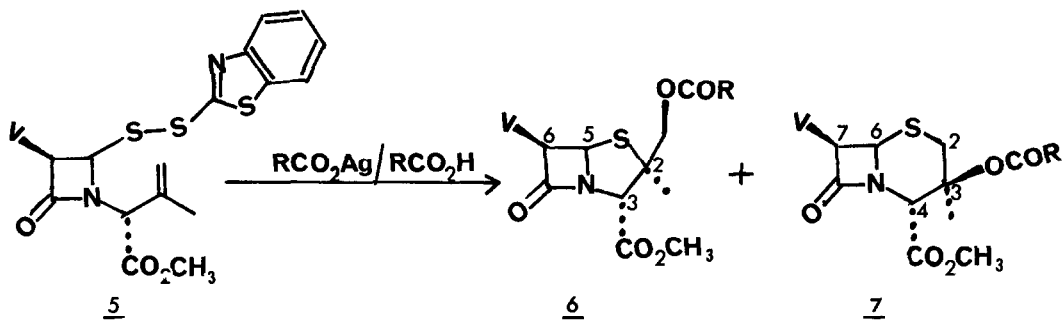
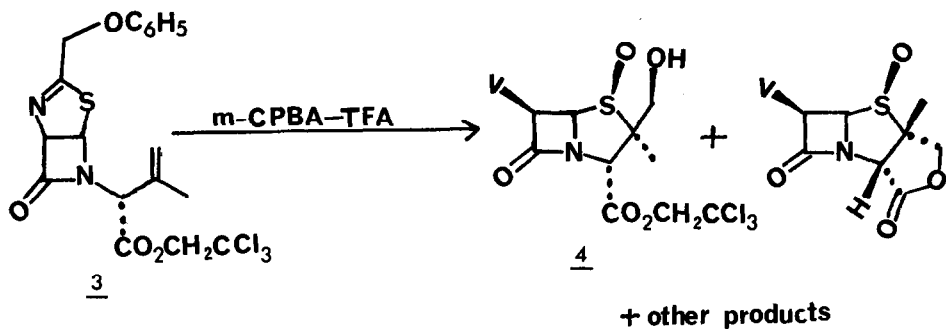
Penicillin derivatives (6a-c) and cephalosporin derivatives (7a-c) were obtained in the ratio given in Table 1 when the disulfide 5 was treated with silver salts of o-nitrophenyl acetic acid, o-nitrocinnamic acid and chloroacetic acid respectively. The reduction of penicillin derivatives (6a & b) with 10% Pd-C resulted in the corresponding amino derivatives (in 6b, the double bond



1a R=H
1b R=OH



2a R=H
2b R=OH
2c R=OCOCH₃



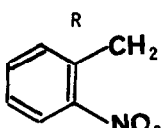
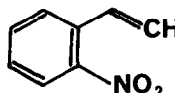
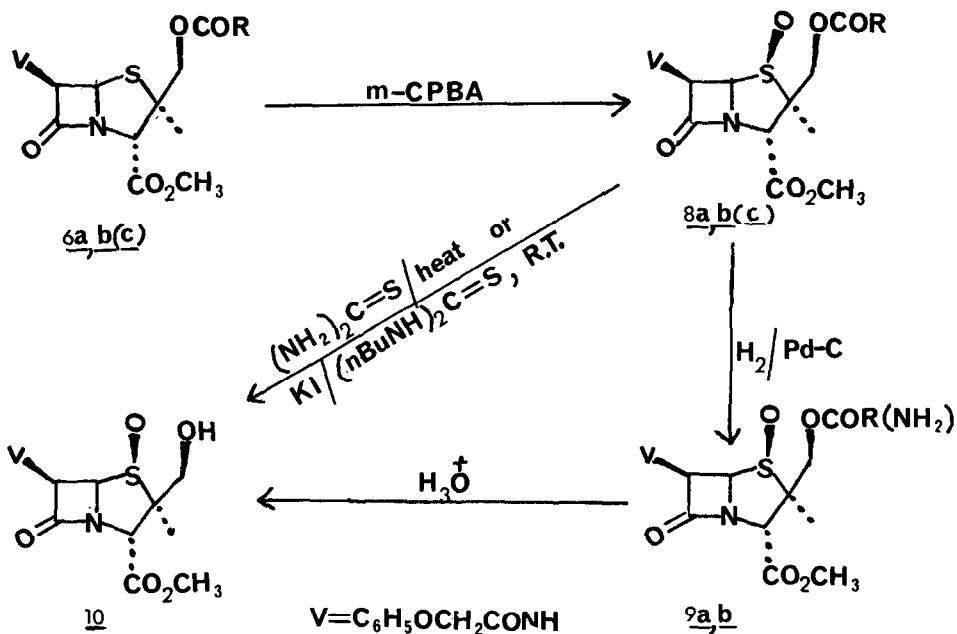
	Reaction condition	Ratio (<u>6</u> : <u>7</u>)	Overall Yield (%)
a	 Benzene, R.T., 38 hrs.	2:1	60.8
b	 DMF, R.T., 70 hrs.	1.5:1:2 (<u>7</u> , R=H)	59.5
c	ClCH₂ Benzene, R.T., 17 hrs. or CH ₂ Cl ₂ , R.T., 4 hrs.	1:1	70.2

Table 1



Comparison of chemical shifts in 6 and 7

	Ring CH_3	$2\beta\text{-CH}_2(\text{H}_6)/\text{H}_2(7)$	$\text{H}_3(6)/\text{H}_4(7)$	$\text{H}_5, \text{H}_6(6)/\text{H}_6(7)$	$\text{H}_7(7)$
<u>6a</u>	1.43	4.62, 4.98 (ABq) $J_{AB} = 12\text{Hz}$	4.67	5.63-5.8 (m)	
<u>7a</u>	1.50	3.40 (s)	4.75	5.3 (d) $J = 4\text{Hz}$	5.7 (d of d) $J = 4, 10\text{Hz}$
<u>6b</u>	1.53	3.93, 4.53 (ABq) $J_{AB} = 11\text{Hz}$	4.8	5.53-5.8 (m)	
<u>7b</u>	1.65	3.55 (s)	4.8	5.3 (d) $J = 4\text{Hz}$	5.6 (d of d) $J = 4, 10\text{Hz}$
<u>6c</u>	1.50	4.01, 4.3 (ABq) $J_{AB} = 11\text{Hz}$	4.66	5.56-5.8 (m)	
<u>7c</u>	1.60	3.46 (s)	4.8	5.33 (d) $J = 4\text{Hz}$	5.70 (d of d) $J = 4, 10\text{Hz}$

Table 2

is also reduced), which upon treatment with acid yielded only complex mixtures. The penicillin derivatives 6a, b & c were oxidised to the corresponding sulfoxides 8a, b & c in 86.3, 94.2 and 75.8% yield respectively. Compounds 8a & b on hydrogenation with 10% Pd-C afforded the corresponding amino derivatives 9a & b (the double bond in 8b is also reduced) in quantitative yields. Subsequent treatment with dil. HCl yielded 2- β -hydroxymethyl penicillin V sulfoxide methyl ester (10) in 49.4 and 68.2% yield respectively [n.m.r. (CDCl₃): 1.23 (s, 2-CH₃), 3.86, 4.16 (ABq, CH₂-O), 5.0 (s, H₃), 5.03 (d, 4Hz, H₅), 6.16 (d of d, 4Hz, 10Hz, H₆); i.r. (CHCl₃): 1800, 1745, 1690 cm⁻¹; mass spectrum M⁺ = 396]. Since we have established the instability of 2- β -hydroxymethyl penicillin, a milder method for deprotection was sought. 2- β -Chloroacetoxyethyl penicillin sulphoxide 8c on heating with thiourea in absolute ethanol gave the compound 10 in 70% yield after chromatography.¹⁰ The same conversion was more conveniently achieved by treating penicillin derivative 8c with KI followed by N,N'-di-n-butylthiourea in acetone at R.T. in 74% yield.

We are presently investigating the application of the above procedure to the synthesis of 2- β -hydroxymethyl penicillin N.

Acknowledgments

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